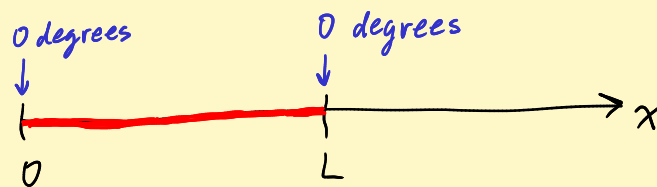


Lecture 18 The heat equation

1



$$\frac{\partial u}{\partial t} = c \frac{\partial^2 u}{\partial x^2} \quad \text{PDE}$$

$$\begin{cases} u(0,t) = 0 \\ u(L,t) = 0 \end{cases} \quad \text{B.C.}$$

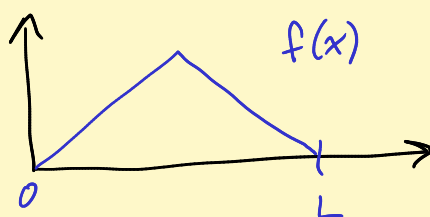
$$u(x,0) = f(x) \quad \text{I.C.}$$

↑ given

$$u(x,t) = \sum_{n=1}^{\infty} B_n e^{-c \left(\frac{n\pi}{L}\right)^2 t} \sin \frac{n\pi x}{L}$$

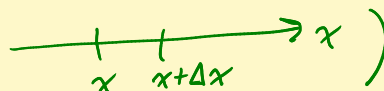
where $B_n = \frac{2}{L} \int_0^L f(x) \sin \frac{n\pi x}{L} dx$

Two problems

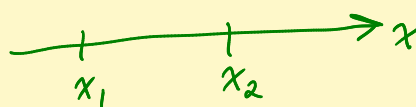


See Maple demo

How to derive the heat eqn (Stein does the infinitesimal version:



"Control volume" method.



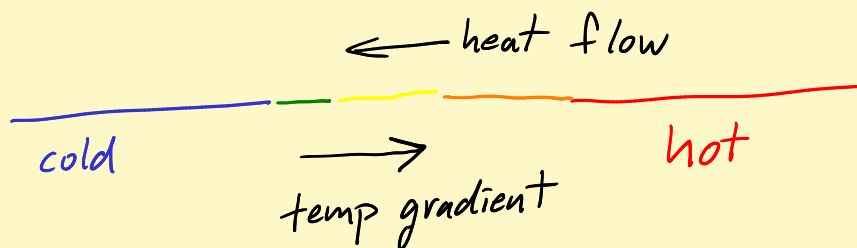
1 calorie is the heat energy required to raise one gram of water one degree Celcius (= kelvin).

$$\text{Heat in } (x_1, x_2) = K \int_{x_1}^{x_2} u(x,t) dx$$

2

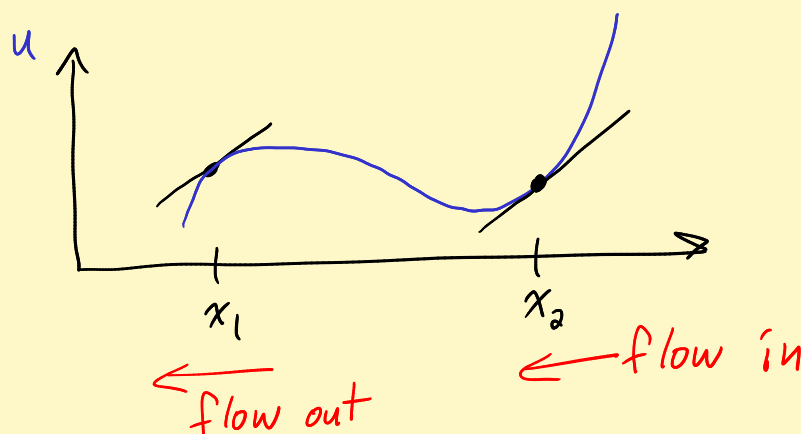
$$\text{Rate it changes} = \frac{\partial}{\partial t} \left[k \int_{x_1}^{x_2} u(x,t) dx \right] = k \int_{x_1}^{x_2} \frac{\partial u}{\partial t}(x,t) dx$$

Kelvin's law



$$(\text{Rate of heat flow}) = -c (\text{temp gradient})$$

EX

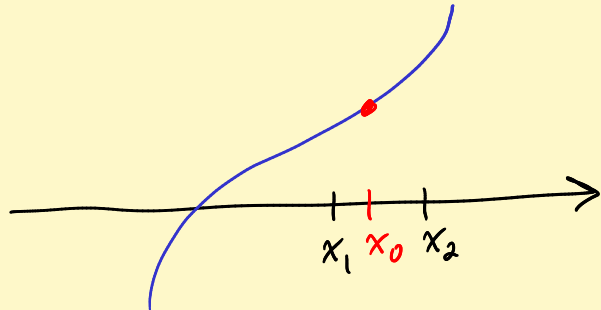


$$\left(\text{Rate of change of heat in } (x_1, x_2) \right) = k \int_{x_1}^{x_2} \frac{\partial u}{\partial t}(x,t) dx = \underbrace{c \frac{\partial u}{\partial x}(x_2,t) - c \frac{\partial u}{\partial x}(x_1,t)}_{\substack{c \int_{x_1}^{x_2} \frac{\partial^2 u}{\partial x^2}(x,t) dx \\ \text{Fund Thm Calc}}}$$

So

$$\int_{x_1}^{x_2} \left[\underbrace{k \frac{\partial u}{\partial t} - c \frac{\partial^2 u}{\partial x^2}}_{q(x,t)} \right] dx = 0$$

Aha! x_1 and x_2 can be anything!



$$\varphi(x_0, t_0) > 0$$

$$\text{Get } \int_{x_1}^{x_2} \neq 0$$

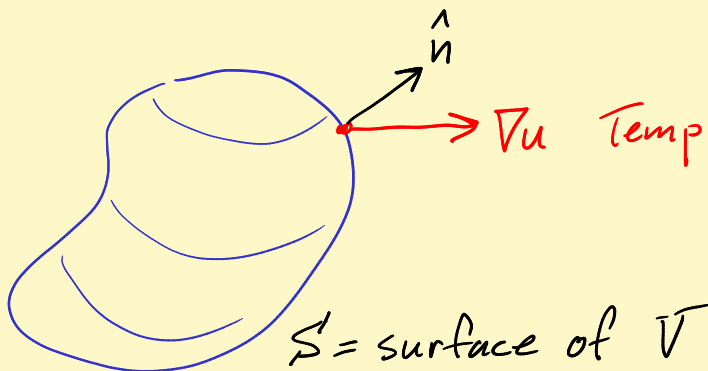
Conclude $[] \equiv 0$

$$\frac{\partial u}{\partial t} = \frac{c}{k} \frac{\partial^2 u}{\partial x^2}$$

3-D version

Control
volume

V



$$\left(\text{Rate of change of heat in } \bar{V} \right) = \frac{\partial}{\partial t} \left(k \iiint_{\bar{V}} u(x, t) dV_x \right)$$

= Net heat flow into $\bar{V}$

$$= c \iint_S \nabla u \cdot \hat{n} \underbrace{d\sigma}_{\text{area}} \quad \leftarrow \text{Flux of } \nabla u$$

$$\stackrel{\text{Div Thm}}{\equiv} c \iiint_V \underbrace{\text{Div}(\nabla u)}_{\Delta u} dV$$

$$0 = \iiint_V \left[\underbrace{k \frac{\partial u}{\partial t} - c \Delta u}_{\text{must } \equiv 0 \text{ because } \bar{V} \text{ arbitrary.}} \right] dV$$

Other heat problems

1) Homog boundary conditions: Slap on refrigerators.

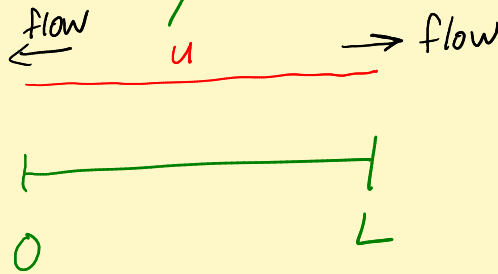
2) Insulated end: No heat flowing. Kelvin $\Rightarrow$

Temp gradient = 0 : $\frac{\partial u}{\partial x}(L, t) = 0$ for all t .
 $\uparrow$ right end insulated.

3) Radiative boundary condition

(u in Kelvin is always ≥ 0)

$$\frac{\partial u}{\partial x}(0, t) = c u(0, t)$$



$$\frac{\partial u}{\partial x}(L, t) = -c u(L, t)$$